

# COMPOSITION

MODERN

# BOOK

[www.kurtzbros.com](http://www.kurtzbros.com)

NAME Michael Plasmeier  
SUBJECT Science Log PD 3 GRADE 3

HR 130



Michael Plasmeier

1. I think space (not air, which is made of particles) is between the particles
2. They visualize it being made of particles because it must be made of something. They also have high powered microscopes.
3. Doesn't the water evaporate and turn to vapor and float away (did this last year)

## Going Further Qu

1/3

1. What will you end up with if you keep breaking a rock into smaller pieces?

You will end up with particles → then → atoms → quirks,

\* Compound → made of more than 1 element

HW

# Observations? Inferences

1/3

1. I see (observe) ice and can infer that it's cold to the touch.
2. I see water and can infer that it's wet.
3. I see the stove is orange so it must be hot.
4. I smell spaghetti sauce, so spaghetti must be for dinner
5. I saw my dad come in with his coat wet, so it is probably raining.
6. I hear pitter-patter on the roof, so it is probably raining.

# Particle Theory + Notes

1/6

All matter is made of small pieces. (particles)

Notes

1/6

1. A particle model of matter can be used to explain observations of dissolving, pouring and mixing substances.
2. A case favoring the particle nature can be made based on observation and inferences

# Lesson 2: Hidden Structure

## Notes

1/18

1. All matter is made of atoms.
2. There is a finite number of elements that occur naturally on Earth (92)
3. Atoms of the same element have the same properties
4. Atoms combine to form molecules.

### Terms:

- atom - the smallest particle into which an element may be divided and still be the same substance
- compound - a substance consisting of two or more elements that are chemically combined
- element - a substance that consists of only one type of atom and that cannot be separated into other substances by ordinary chemical changes
- molecule - the smallest unit of a substance that has all of the physical and chemical properties of the substance and that is composed of 2 or more atoms

# Concept Map

John Dalton

1/20

Observation

When passing electric current through water the elements are separated  
water can be separated into 2 elements which are always recovered in the same proportion

Question

What could be inferred from this?

Inference

The particles of different elements have different masses because the mass of Oxygen in water is different from the mass of hydrogen in water.

Explanation

The particles of atoms or a single element have identical masses and properties because the same ratio of oxygen to hydrogen was obtained every time. An atom of element can not be charged into an atom because 2 different substances were found to make up water.

Model

The balls of modelling clay represent atom



1. The model oxygen atom is 8x heavier than the model hydrogen alone supporting his theory
2. The mass is always 8x greater
3. There will always be 2x more than hydrogen

# Testing Understanding

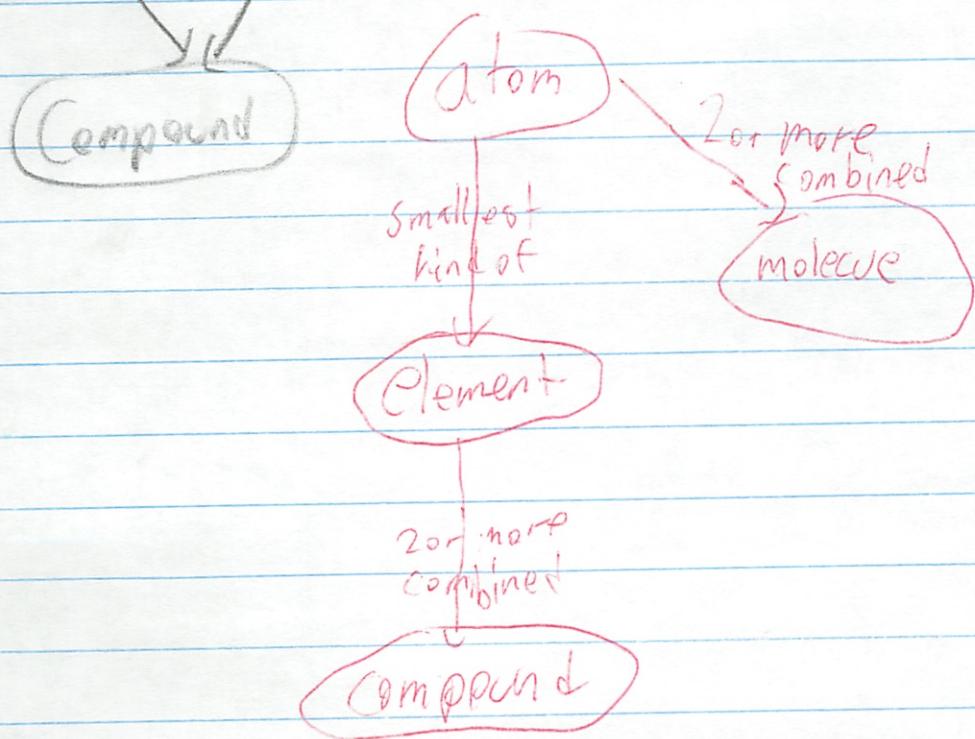
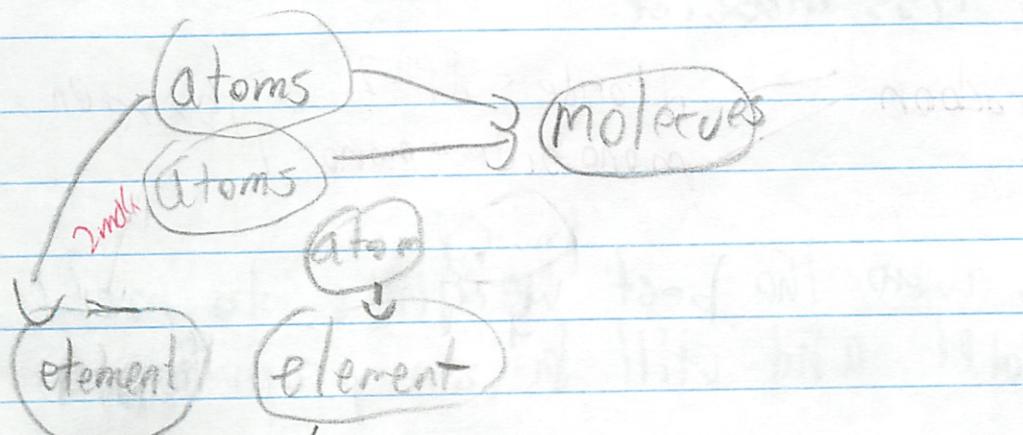
p 96

1/26

1.

2 like make element

2 unlike make compound



other way around

2. a  $\text{CO}_2$  is less massive <sup>in carbon monoxide</sup>
- b. There is 1 less oxygen so it would be less massive
- (Carbon)  $\rightarrow$  dioxide = di - 2      ) oxygen  
 monoxide = mono = 1

3. Those were the best weights to make a model and still be in proportion.

# Particles in Motion

## Summary

2/1

1. Joseph Black studied the effects of heat on matter.
2. Cohesion <sup>is the</sup> when molecules attach to like particles
3. Adhesion is force of attraction between unlike particles
4. The surface of water forms a curve called a meniscus
5. Water adheres to glass to move up the side of a glass
6. Surface tension hold the molecules in a drop and lets insects walk on water.
7. The kinetic theory of matter says that the particles are in constant motion.
8. Solids have a orderly fixed arrangement

of molecules

9. Liquid molecules can move around but are always in contact w/ other molecules,
10. Gas molecules have a lot of kinetic energy and move freely from place to place
11. In a solid the particles don't have enough kinetic energy to break free.
12. If they get enough energy they will start to move
13. Diffusion is the mixing of particles because of the mixing of particles,
14. Diffusion occurs quickly in gasses because the particles can move around quickly.

15. Particles move faster at higher temperature, so diffusion is more rapid at higher temps.

16. liquid stays at 100°C even when heated faster

17. This can be explained because the particles can not move faster without breaking out of cohesion that holds them together.

18. Ice melts when the ice particles gain enough energy to move freely.

19. When substances cool down, it's particles loose kinetic energy and slow down.

20. The cohesion pulls them together and it condenses to a solid.

# In 3 Dimensions

p 98

2/2

1. They are all round and bond together not with sticks. The different atoms seemed to be spaced evenly out.  
all made of atoms biggest in center

They are different because there are different atoms of different sizes and colors. There are also different numbers of atoms different kinds of atoms.

2. 3, 4, 5 There are <sup>about</sup> 87 elements not shown and 3 shown.
3. No, they bond like this

# The Size of Particles Sugar and Starch Molecules

2/9

- Sugar (dextrose) and <sup>(cold)</sup> Starch are made of carbon, hydrogen, and oxygen compounds
- Iodine Solution is a chemical indicator. It is used to indicate or show the presence of starch by turning the sample inky blue to black
- Benedict's solution is a chemical indicator. It indicates or shows the presence of sugar by turning a heated sample bluish, greenish, yellow.

(Data)

## Exploration 2

### Sugar and Starch Molecules

2/9

Period 3

(Properties)

Dextrose (Sugar)

- white
- powdery
- granular
- has small grains
- looks like smell
- no odor

Corn-starch

- wet looking
- white
- powdery
- lumpy
- no odor
- soft

(Results)

No corn-starch molecules  
moved through the membrane

## Summary Ex 2

### size of Sugar and starch Molecules

2/10

1. The molecules of different substances are different sizes
2. Atoms and molecules are extremely small.
3. Some materials allow certain molecules to pass through while blocking others
4. Sugar molecules are smaller than starch molecules
5. Starch turns blue-black in the presence of iodine, while sugar does not.
6. Sugar turns yellow in the presence of Benedict's solution, while starch does not.

- This page left  
blank -

# More Ideas

play

2/17

1. Particles in gasses are too apart.
2. Particles that make up solids and liquids must be as close together as possible.
3. Particles move.
4. Particles in hot substances move faster than particles in cold substances.
5. The faster gas particles move, the more pressure they exert on a ballon.
6. Liquid particles can become gas particles, and gas particles can become liquid particles.

# Picturing An Atom

127-128

2/24

1. John Dalton proposed that matter is made of tiny, invisible particles called atoms.
2. J.J. Thomson conducted experiments that led his "plum pudding" model, a refinement of Dalton's original particle theory.
3. Current model: The atom is a hazy region consisting of a central nucleus surrounded by moving electrons. Atoms differ from one another in the number of protons contained in the nucleus. Gold atoms contain 79 protons while oxygen atoms contain 16 protons.
4. Atoms do not have definite shapes.
5. Atoms do contain identifiable parts (subatomic particles)

(Cont -)

## Picturing an Atom

127-129

p2

2/24

6. Dalton's model represents the atom as indivisible.  
Thomson's model shows the atom contains smaller parts (electrons).

### Reflection p. 28

1. What are main characteristics of an atom as John Dalton?

- indivisible - tiny hard - spherical.

Each has own mass.

John + JJ  
Script

2/24

Thompson: You know a lot has changed in  
the years since you died

Dalton: Well what happened?

Thompson: My experiments shows me that  
atoms are more like pudding than a  
sphere.

Dalton: Why?

Thompson: Atoms have protons and neutrons in their  
nucleus. Electrons are spinning around the outside.

Dalton: So there are things smaller then atoms?

Thompson: Yes, and also atoms aren't completely  
round.

Dalton: There not?

Pudding = soft

ball = hard

# Measuring in Science

3/3

Need units to measure

Everything is measured  
know things will fit

Use instruments to provide measurement  
- lets you measure

North Pole  $\rightarrow$  Equator 10 million m apart  
7 metric tons

$$\text{Area} = l \times w$$

$$\text{Volume} = l \times w \times d$$

a volume of  $1\text{m} \times 1\text{m} \times 1\text{m}$  is 1L of water

1L of water weighs 1 kg

au = 150 million km (astronomical unit)

300 thousand km a second - light = 1 sec

9.5 trillion 2 in light year

race times are measured to hundredths of a second

thermometers measure temp - w/ mercury

# Conservation of Matter

## Conclusion

3/7

1. What invisible substance was in the funnel when you first placed it in the water? Air

2a. When you placed finger over the funnel and put it in the water, no water  $\rightarrow$  to funnel.

The air takes up place in the funnel - can't escape. Water can't coexist w/ air in same place

3/8

2b. How would you explain what happens when you remove your finger.

The water pressure pushes the air out of the top of the funnel

3a Why does the balloon inflate?

The water pushed the air up as before and the air was caught in the balloon.

## Conservation of Matter-Cont

3/8

- 3 b When you squeeze balloon, air is forced out of the balloon and the air pushes the water out of the funnel.

# Burning Race

Notes

3/10

1. Oxidation occurs when element(s) combine w/ oxygen to form compounds
2. 2 familiar oxidation reactions are Combusting (burning) and rusting.
3. Combustion is an oxidation reaction in which oxygen combines rapidly with carbon to form carbon dioxide.
4. Typically, the oxygen in oxidation reactions is taken out of the air.
5. Fanning a fire makes it burn more rapidly by increasing the air flow and therefore the amount of oxygen available in the immediate area around a fire. The fact that combustion is an oxidation reaction also makes it possible to put out a fire by covering it.

## Burning Base

cont

3/10

6. Oxygen makes up approx 20% of the total volume of the air. Oxygen is a very reactive element and forms compounds w/ nearly every other element. Compounds that include oxygen make almost 50% of the earth's crust.

(Conclusion on pg.

# Solution + Mixture Notes

3/10

Homogeneous - mixture that is uniform throughout.  
The composition is the same in all directions in the substance. The types of particles observed in 1 direction are <sup>the same</sup> that are observed in all others. A solution of sugar in water is homogeneous.

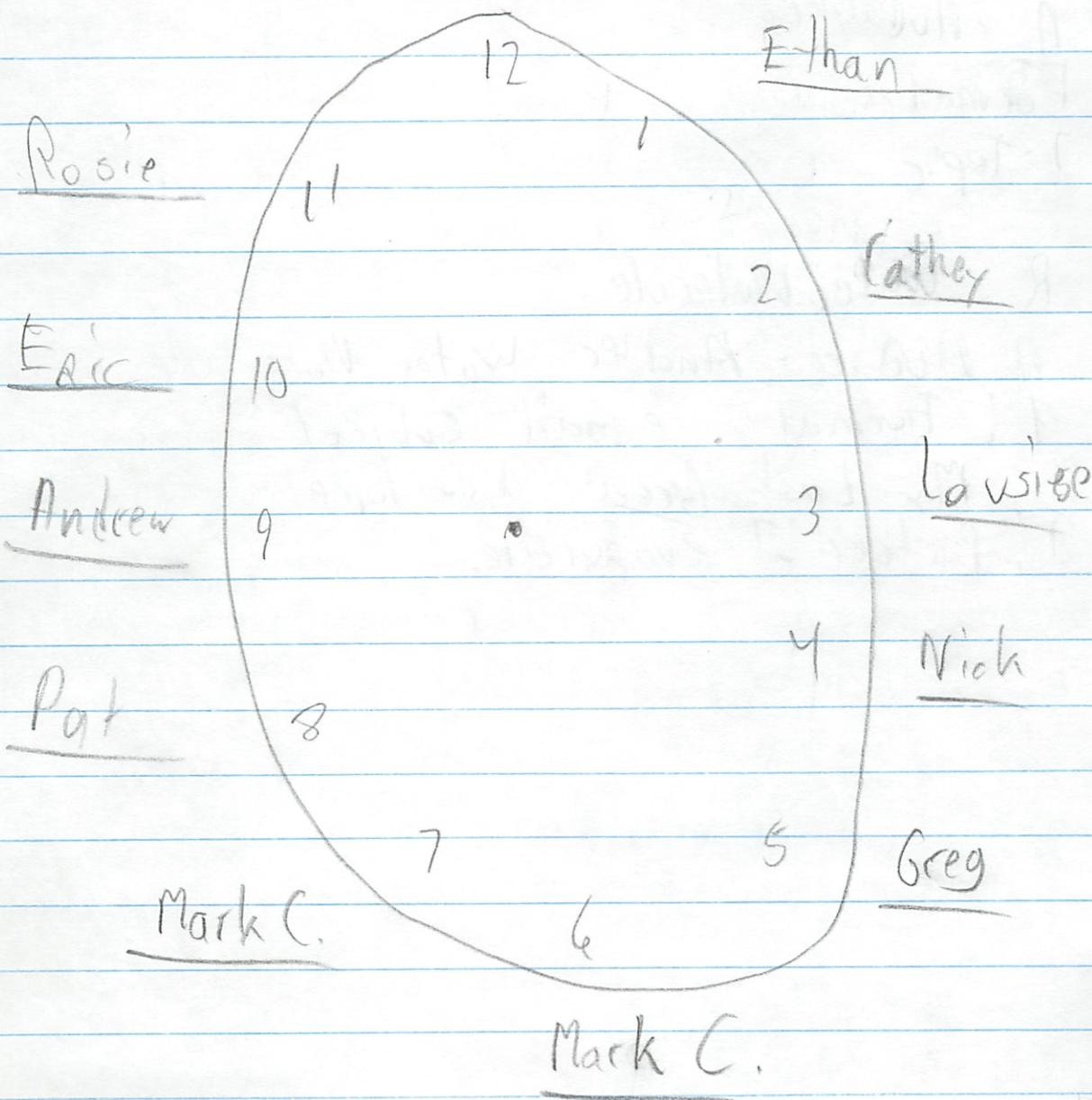
(Opposite is ~~mixt~~)

Heterogeneous - mixtures that are not uniform. There are pockets of one substance surrounded by pockets of different substances. A mixture of soil in water to make mud is heterogeneous.

Partners

3/14

Jordan T.



# Raft Instructions

Due Wed 3/16

R - Role

A - Audience

F - Format

T - Topic

R - Water Molecule

A Audience - Another Water Molecule

F Format - e-mail Subject:

My Last Great Adventure

T Before I evaporate.

# Raft E-mail

3/15

From: Water Molecule #20,759,841,375,942  
To: Water Molecule #1975,481,769,853,321,754  
Subject: My Last Great Adventure #3571,075

Message: Hi, Did you hear about my last adventure before I evaporated. In fact this is my 3,571,075 time that I will make the great journey of being evaporated. Well after my last journey, #3571,074, I condensed in a cloud. I then rained down out of the cloud. I was falling and falling through the blue sky. Splat I finally landed on the roof of my house. I was frantically looking for you, my friend, but I couldn't find you. However before long, I was surrounded by millions of other molecules in a drop of water. We rodded what seemed

Raft cont

3/15

like forever down the side of the roof.  
We clanged into a gutter and we  
began to roll again. This time we were  
getting swept along the swift current  
to the gigantic hole in the gutter. Finally  
our turn came to go down the gutter.  
We were swept down and we took a wild  
ride downward. It was almost as thrilling  
as my last big adventure # 1,875,964  
when I was stuck on a roller coaster ride  
at the theme park. Well anyway, soon  
I ~~clashed~~ fell out of the gutter and  
rolled into a puddle. For a little while  
I was joined by many more molecules,  
but soon it stopped raining. It was a warm  
day so soon it was time for me to make  
the great journey once again.  
of evaporation

Regards # 20,759,541, 375, 947

# Notes

## Exp. Design

1. For the dependent variable you can use quantitative or qualitative data:
  - a. quantitative include measurements w/ standard scales, such as  $^{\circ}\text{C}$
  - b. qualitative include verbal description or measurement w/ non-standard scales for example, the color or clarity

1. IV  
2. DV

3. Rep Trials

Exp Design

Cards

3/29

4. C  
5. Control

Gloria 1. color of mashed potatoes

2. <sup>not</sup> students choosing

3. 100

4. same bowl

same food - type (potato, instant)

same school

scoop size

could change

same age kids

order of bowls

5. home (should use white potatoes)

type of food

Susie 1. Height of the hole

2. How far the water squirts

3. 1

4. Same cartons

same size holes

same amt of water

5. None (no hole)

Improvements

do more trials

specify liquid (type, temp) and specify container, hole shape

have a control



cont

Sandy 1. amt of seeds

2. which ways it looks e best

3. same cups      same mixture of  
same mixtures of seeds      soil  
same time periods      amt water

4. 1

5. None (I did plant multiple cups)

Imp use 1 type of seed  
control  
define cup  
better DV - specify  
light, temp, water  
procedure

Ester 1. type of insulation

2. temp of water in cup

3. 1

4. same jar & insulation  
Same lids, amount

Same type of water

amt of sunlight      same lids

Same amt of  
insulation

5. None (Should have jar w/ no insulation)

Improvement  
add control

Add repeated trials

Specify shape, color of jars

initial temp

time in sunlight

Cont →

# Exp Design

Cards cont

- Pete's
1. amt of distilled water in ml
  2. height of plant should have defined
  3. 10 ~~10~~ amt of seeds types of pots
  4. same type of water - same amt of soil  
Should have same sunlight time watering freq.
  5. pot w/ recommended amt of water (150ml)

Herr

## Improvements

type of seed (age, brand, species)

pots must be identical

Specify light, temp, location

make more measurements of DV

define how to measure height

## Writing Data Pairs

4/1

Points on a graph are represented by a set of data (ordered pair). The values for horizontal (x) axis is written 1st, followed by corresponding value for (y) vertical axis.

$$(10, 48\frac{1}{2})$$

The 2 numbers (a pair) are separated by a comma. Both values are placed in parentheses.

## How did it happen Tsunami;

On Dec 26, 2004 at a subduction zone, a piece of the Eurasian plate rushed upwards and displaced a lot of water quickly. The water shot upwards and gravity pulled it back down forcing it to expand out in all directions. It travels as a huge wave expanding faster and faster till it broke on the coast of South Asia. The massive wave flooded many homes and killed many people. 280,000 people died because they weren't warned. Could this have been prevented?

# Determining Scales for Axis

4/6

X - take largest value - smallest #

then / 5 and round to convenient #

$$\text{ex } 40 - 10 = 30 / 5 = 6 \text{ round to 5}$$

Do same to Y

# Summarizing Trends

4/6

Because experimental data are subject to error, data points on a graph are not directly connected. Instead a line-of-best fit that communicates the general data pattern used.

To construct a line-of-best fit draw a line about which an equal number of points fall to either side.

\* Underneath graph write a summary statement:  
As the paper towel is submerged in a ~~water~~ water for longer, the water-level will rise.

~~Investigating Human Traits~~

Invest 54

8/15

How much variation is shown by students in your class

# Questions

## Invest 54

4/26

1. What causes variation among humans that you observe in this activity?

We noticed that some traits occurred more often than others. For example 18 people could roll their tongues, while only 4 couldn't. We think these changes are caused by genes.

2. What patterns do you see in the data for the different characteristics?

Some traits a lot of people have like tongue rolling, and finger rolling. Most people could do both things. We did not take specific measurements for people being able to do both.

3. What characteristics are easier to categorize? Why?

How? In what way. ~~Height~~, finger, and tongue rolling are easier to graph because you can or you can't, there is no in between. Height + Armspan wasn't collected on the graph data table right, so it is hard to graph.

# Invest 54 Final Conclusion Notes

4/27

1. We get traits from our parents or grandparents, and differences in the environment or during development and growth.
2. Nature vs. Nurture or heredity vs. environment is used to distinguish between traits than an individual is born w/, either due to heredity or other factors, and traits that an individual acquires as a result of experiences, such as family experience or education.
3. Skin color and height both have a basis in heredity, but skin coloring can be affected by the amount of sunlight which a person is exposed, and height can be affected by dietary factors.
4. Age and what a person has eaten recently are 2 factors that might affect PTC tasting, which does have a genetic basis.
5. Heredity (or genetics), is the study of how traits are passed from parents to their offspring and subsequent generations.
6. PTC tasting is an all-or-nothing trait while height shows

## 54 Notes cont.

Continuous variations over a long range, and eye color shows 4 or 5 distinct color families, but variation of those colors exist as well. Tongue rolling maybe difficult to categorize.

Invest 55

Prediction

Plants have Genes too

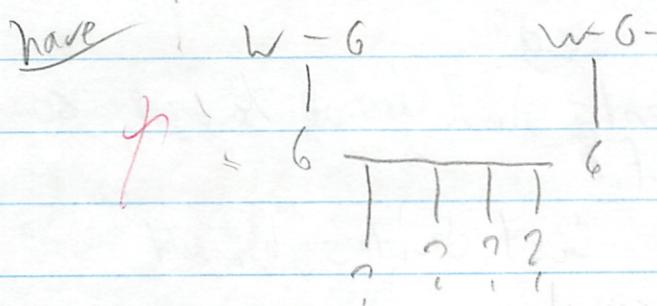
4/27

I think that  $\frac{1}{4}$  of the plants will be white and  $\frac{3}{4}$  would be green. This is because dominant recessive genes. The green trait is dominant so about

P <sub>parent</sub>		F <sub>1</sub>	
W	w	W	w
W	w	W	w
W	w	W	w
W	w	W	w

$\leftarrow \frac{1}{4}$  would have both be white

?



Q) if that is for parents, then will the GG not count  
Parents = GW GW

Yes that is what diagram shows  
so prior conclusion  $\rightarrow$  stands

# Invest 57 Copycat

## Analysis

1. Classify as asexual or sexual reproduction

a. Orange cat w/ black cat

- Would be sexual - 2 parents.

b. cutting from flower in water

asexual - 1 parent

c. w/ geranium bred w/ another

sexual - 2 parents

d. sperm fertilizes an egg

sexual - 2 parents - don't have to make contact

e. worm splits in half.

asexual - 1 parent - just divides himself

f. cloning: cloning is asexual

asexual - DNA is from only 1 parent

2. Would you clone yourself?

That is a big controversy. Do you

clone humans? The technology is not

there today to clone humans. I really

I want to be unique + could be complications

# Invest 56 Marfan Syndrome

## Analysis

1. Signs have Marfan Syndrome.

loose arms + joints bending fingers fall  
long face slender stature easily tired etc curve of spine

2. What causes it? 3/4 cases inherit it spontaneous mutation  
Genes - you inherit it 1/4 cases -> during reproduction

3. Can you catch "it"

No. You need to inherit it from your parents.

4. Can't exercise to hard. May have more frequent heart attacks. (Can choke off to friends) (Effect on your life)  
<sup>Maby lost eyesight</sup>

5b. Question to ask counselor → See Next Page #1

is it dominant or recessive - chance that I would have it - chance kids would have it

6. How would behave towards kid w/ Marfan?

Just the same. He wouldn't be as good as sports, just like me, so just the same.

## Invest \$6 Joes Dilemma

Q:

Would you want to find out if you had a genetic disease?

Yes, so I would know if my kids could get it ~~too~~ (so that way I can avoid dying because if I knew I had a disease like Marfan syndrome I wouldn't play Basketball too hard so I wouldn't have a heart attack and die.)

Have condition → What's causing it

## 56. Sjögren's dilemma

### Questions

1. What questions would you ask before being tested.

- is it dominate or recessive
- what is the chance I would have it
- chance kids would have it could be 50% play soccer
- chance kids have it if married girl w/ Marfan
- how much \$ cost the test? Are there any medicines?
- easier way? If I have it - be further costs?
- is it covered by insurance Do I have to visit hospital?

### 2. Advantages

change lifestyle

make kids not have it

live life to fullest

don't over exert myself

know if medicine comes

along in future

believe in science

know what it is when can't see

### Dissadvantages

\$

can't change it

don't know that you will die

keep to nature

can't play soccer

## 56. Revisit Analysis

5. It seemed like it was normal  
50% chance kid will have it  
b I already seen movie before- so results invalid,  
She says would have more doctor visits
- Do I get out of  
gym class.
6. I could care less... that he is different.  
It's not all that serious of a disease, and in school  
we always are told to accept other's differences.

57 Copycat

Question

5/9

Difference between sexual + asexual

Sexual - 2 parents - traits from both parents (2 unlike cells)

Asexual - 1 parent - (clones) identical from parent

## 57 Copy Cat

Stop to think

5/1

1. If bacteria divide, which is signal.

Hard to tell. One are no mutations or a smaller size. If both have identical DNA to the original, and both same size - can't tell  
Or say that both halves are "children"

2. How do multi-cell organisms reproduce asexually?

They break parts of themselves off or they produce "tunne's" or when you take cuttings

3. Why are identical twins similar than fraternal twins.

The DNA mixes in egg, and egg is the same, so because identical from same mixture = same  
fraternal - Sporin gets 2 eggs and 2 separate mixtures occur

4. How is clone different from identical twin

An identical twin is still a mixture from 2 parents.

A clone is from 1 parent.

## 6.2 Analyzing Garlic Data

5/10

1. Seeds are the result of sexual reproduction

Green	Yellow	Non Sprout
8	3	-

We can distinguish  
between these 3  
seed coats

# 58 Creature Features

Chalange Q

5/11

How are traits passed from parents to offspring.  
Through genes. The sex cell contains half of the chromosomes. If a sperm meets an ova, the zygote has 56 chromosomes, a combination of traits from the mother + father.

Stop to think1 - all of 1 color - whatever color is dominate.

Stop to think2 - Blue is a dominate trait over orange

ST 3 - I was correct

Alt Hyp 58.2 - Picked C - because I know what these investigations are about

## 6.2 Plant Genes

### Follow-Up

1. The grandparents are a normal green flowering tobacco gene (GG) and a purebreed strain of albinos mutants (gg), which do not produce the chlorophyll responsible for the green color of plants.

2. The parents of seeds germinate are the cross between the GG and gg plant

$$\begin{array}{l} \text{GG} \times \text{gg} \\ \text{Gg} \quad \text{Gg} \quad \text{or same} \\ \text{Gg} \quad \text{Gg} \end{array}$$

	<u>G</u>	<u>g</u>
<u>G</u>	<u>Gg</u>	<u>g</u>
	<u>Gg</u>	<u>g</u>

3. The parent offspring (Gg) get the G from the green grand parent and the g from the albino grand parent.

4. Next Generation

$\text{Gg} \times \text{Gg}$	$\text{Gg}$	$\text{gg}$
$\text{Gg}$	$\text{Gg}$	$\text{gg}$
$\text{gg}$	$\text{gg}$	$\text{gg}$

$\frac{1}{4}$  are albino

	6	g
6	66	6g
g	g6	gg

## 62 Follow Up

Coh

5. These parents (Gg) all appear green + totally normal  
(but they have a hidden recessive genes and  
this called a carrier of the pale yellow  
traits (Also called hybrid or heterozygote)  
(See diagram - 1 pg)

6. Since GG and Gg are both green in the  
ratio of  $\frac{3}{4}$  green and  $\frac{1}{4}$  albino (3:1 phenotype)  
the phenotype describes the physical characteristics,  
it is a  $\frac{1}{4}$  GG,  $\frac{1}{2}$  Gg,  $\frac{1}{4}$  gg genotype (1:2:1  
genotype) A genotype is the genetic  
makeup of characteristics. Genotype + environmental  
conditions determine phenotype.

7. ~~Genotype~~ Phenotype

# 60; Mendel's Notes

5/24

Why do we breed dogs, cats, etc...?

People want the best traits in the animals so they hope to combine + breed 2 animals to take the best traits from each.

Breeding ~~of~~ refers to the process of selecting 2 parent individuals to mate (animals) or cross (plants) in order to produce a desired outcome or improvement in the offspring.

(Ques cont)

Stop to think 3; How is S9 like Mendel's plants.

Recons model Mendel's experiment by giving a trait at random from both parents to offspring

Analysis p.38

1. What dominate/recessive traits

See page

Sci Def for  
Dominant

5/24

A dominant trait is a trait that you can always observe if at least 1 allele for the trait is present. For example, if the blue-tail trait is dominant and observed even if an allele for the other trait (orange tail) is present.

## 59: Gene Combos

5/20

♂ Tails - blue tail (T)

♀ Heads - orange tail (t)

Blue tail is dominant to orange tail color

Reminder T = blue tail gene

t = orange tail gene

Analysis

5/23

a. blue: 89  
 ocean: 31  
 $\text{ratio} = \frac{89}{31} = 2.87$

b. 3:1

c. About  $\frac{3}{4}$  (3:1) of the offspring have blue tails and  $\frac{1}{4}$  (1:3) have orange tails

d. Blue is dominant / ~~H/T~~  
 3 combos for blue, 1 for orange

e. a-d = 25

H	H	H	H
t	t	t	t

f. No, it's just the expected probability, actual is different

g. Yes, because the coin-tossing gives  $\frac{1}{2}$  chance like a parent giving a gene at random

## 60: Mendel

5/19

Stop to Think 1: What qualities did Mendel have?

He must have had patience, persistence, good observation skills. He could not have been too shy to share his findings. He had to be good at science + math

Stop to Think 2: a. advantages to using pea plants

- grow quickly
- make lots of seeds
- simple traits that come in 2 varieties + don't blend

Stop to Think 2b: Why so many crosses?

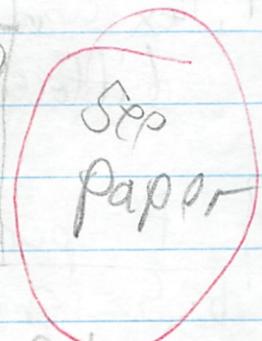
To have a high # of repeated trials

(cont.)

68. Merkel

5/24

Flour	Seed	Soil	Plant
White	green vs yellow	winter	green vs yellow
Purple	yellow	Smooth	green
White	yellow	Smooth	



2 b. Again actual vs. theoretical - error factors may contribute

ratio are another  $\frac{\text{Part}}{\text{whole}}$

3. 1:3 means it happens 1 time over (1+3) times etc.

4. More samples, etc - to get more conclusive data

5. Math is important in every science and also other subjects

# GL Analysis

1. They were both written to be similar, and it's a different way to do the same thing (I chart, I flip)  
[I hero, I actual]
2. F = purple f = white  
b. S = smooth s = wrinkled
3. The dominant trait is stronger or darker.

4. Cross a few purple w/ white or look at family tree to see if white offspring come 1/2 the time

If so, then hetero.

		Purple		White	
		P	p	p	P (p)
white (p)	P	P(p)	P(p)	P(p)	P(p)
	p	P(p)	P(p)	P(p)	P(Hh) (pp)

all hetero - purple

2 hetero - purple

2 homo - white

## 63: Show Genes

Stop to think  
Questions

5/27

StoT1 - Cell division in single cell org. is asexual reproduction. In multi-celled org it is to create new cells to replace or add to old ones.

StoT2 - The # of chromosomes would keep being cut in half till none are left

StoT3 - That way when they join, there would be 46 not  $^{47}$

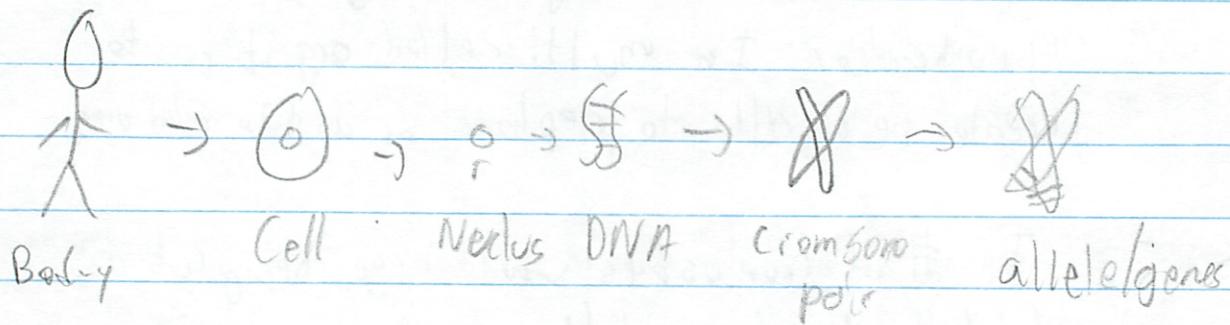
StoT4 - Which chromosome is given is random and also enviro factors, and multiple genes for a trait

StoT5 Mutations occur when the DNA is copies and a mistake occurs

Analysis 1

Next pg

## 63. Analysis



A person's body contains many cells. In the middle of each cell is a nucleus. Inside is the DNA. This is made of 23 chromosome pairs which have many genes on them.

# Useful Information

MULTIPLICATION TABLE

<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b>	<b>10</b>	<b>11</b>	<b>12</b>
<b>2</b>	4	6	8	10	12	14	16	18	20	22	24
<b>3</b>	6	9	12	15	18	21	24	27	30	33	36
<b>4</b>	8	12	16	20	24	28	32	36	40	44	48
<b>5</b>	10	15	20	25	30	35	40	45	50	55	60
<b>6</b>	12	18	24	30	36	42	48	54	60	66	72
<b>7</b>	14	21	28	35	42	49	56	63	70	77	84
<b>8</b>	16	24	32	40	48	56	64	72	80	88	96
<b>9</b>	18	27	36	45	54	63	72	81	90	99	108
<b>10</b>	20	30	40	50	60	70	80	90	100	110	120
<b>11</b>	22	33	44	55	66	77	88	99	110	121	132
<b>12</b>	24	36	48	60	72	84	96	108	120	132	144

## METRIC MEASURE

### Measures of Length

10 millimeters	= 1 centimeter, cm
10 centimeters	= 1 decimeter, dm
10 decimeters	= 1 meter, m
10 meters	= 1 decameter, dam
10 decameters	= 1 hectometer, hm
10 hectometers	= 1 kilometer, km
10 kilometers	= 1 myriameter, mym

### Measures of Weight

10 milligrams, mg	= 1 centigram, cg
10 centigrams	= 1 decigram, dg
10 decigrams	= 1 gram, g
10 grams	= 1 decagram, dag
10 decagrams	= 1 hectogram, hg
10 hectograms	= 1 kilogram, kg
10 kilograms	= 1 myriagram, myg
10 myriagrams	= 1 quintal, q
10 quintals	= 1 millier or ton, MT or t

### Measures of Capacity or Volume

10 milliliters, ml	= 1 centiliter, cl
10 centiliters	= 1 deciliter, dl
10 deciliters	= 1 liter, l
10 liters	= 1 decaliter, dal
10 decaliters	= 1 hectoliter, hl
10 hectoliters	= 1 kiloliter, kl

1 liter = 1 cubic decimeter;

1 kiloliter = 1 cubic meter

## CONVERSION TABLE

When You Know:	You Can Find:	If You Multiply By:
Inches	Millimeters	25.4000
Feet	Meters	.3048
Meters	Feet	3.2809
Yards	Meters	.9144
Miles	Kilometers	1.6093
Kilometers	Miles	.6213
Sq. Inches	Sq. Centimeters	6.4515
Sq. Feet	Sq. Meters	.0929
Acres	Hectares	.4047
Cu. Inches	Cu. Centimeters	16.3866
Cu. Feet	Cu. Meters	.0283
Cu. Yards	Cu. Meters	.7650
U.S. Gallons	Liters	3.7854
Quarts	Liters	.9460
Pints	Liters	.4730
Ounces	Grams	28.3495
Pounds	Kilograms	.4536
Tons	Metric Tons	.9072

## TEMPERATURE

Under the Metric System, temperature is recorded in degrees Celsius instead of Fahrenheit.

Some ideas of the difference between these can be realized if you consider room temperature of about 68° Fahrenheit, would be 20° Celsius. Water would freeze at 0° Celsius, which is 32° Fahrenheit.

## U.S. MEASURE

### Long Measure

12 inches	make 1 foot
3 feet	make 1 yard
6 feet	make 1 fathom
5½ yards	make 1 pole or rod
40 poles	make 1 furlong
8 furlgs	make 1 mile
69⅔ miles	make 1 degree
320 rods	make 1 mile

### Miscellaneous

12 units	make 1 dozen
12 dozen	make 1 gross
20 units	make 1 score

### Solid or Cubic Measure

1728 cu. in.	make 1 cu. foot
27 cu. ft.	make 1 cu. yard
128 cu. ft.	make 1 cd. wood
24⅓ cu. ft.	make 1 perch stone

Note: - A cord of wood is a pile 8 ft. long, 4 ft. wide and 4 ft. high; therefore, 8 ft. x 4 ft. x 4 ft. = 128 cu. ft.

A perch of stone or brick is 16½ ft. long, 1½ ft. wide and 1 ft. high.

### Liquid Measure

2 cups	make 1 pint
2 pints	make 1 quart
4 quarts	make 1 gallon
3½ gallons	make 1 barrel

63 gallons	make 1 hogshead
144 sq. in.	make 1 sq. ft.
9 sq. ft.	make 1 sq. yd.
30¼ sq. yds.	make 1 sq. pole
40 sq. poles	make 1 rood
4 roods	make 1 acre
640 acres	make 1 sq. mile